



## Strategies to increase CO<sub>2</sub> solubility

**BioCON-CO<sub>2</sub> – Final Symposium**

**13.06.2022**

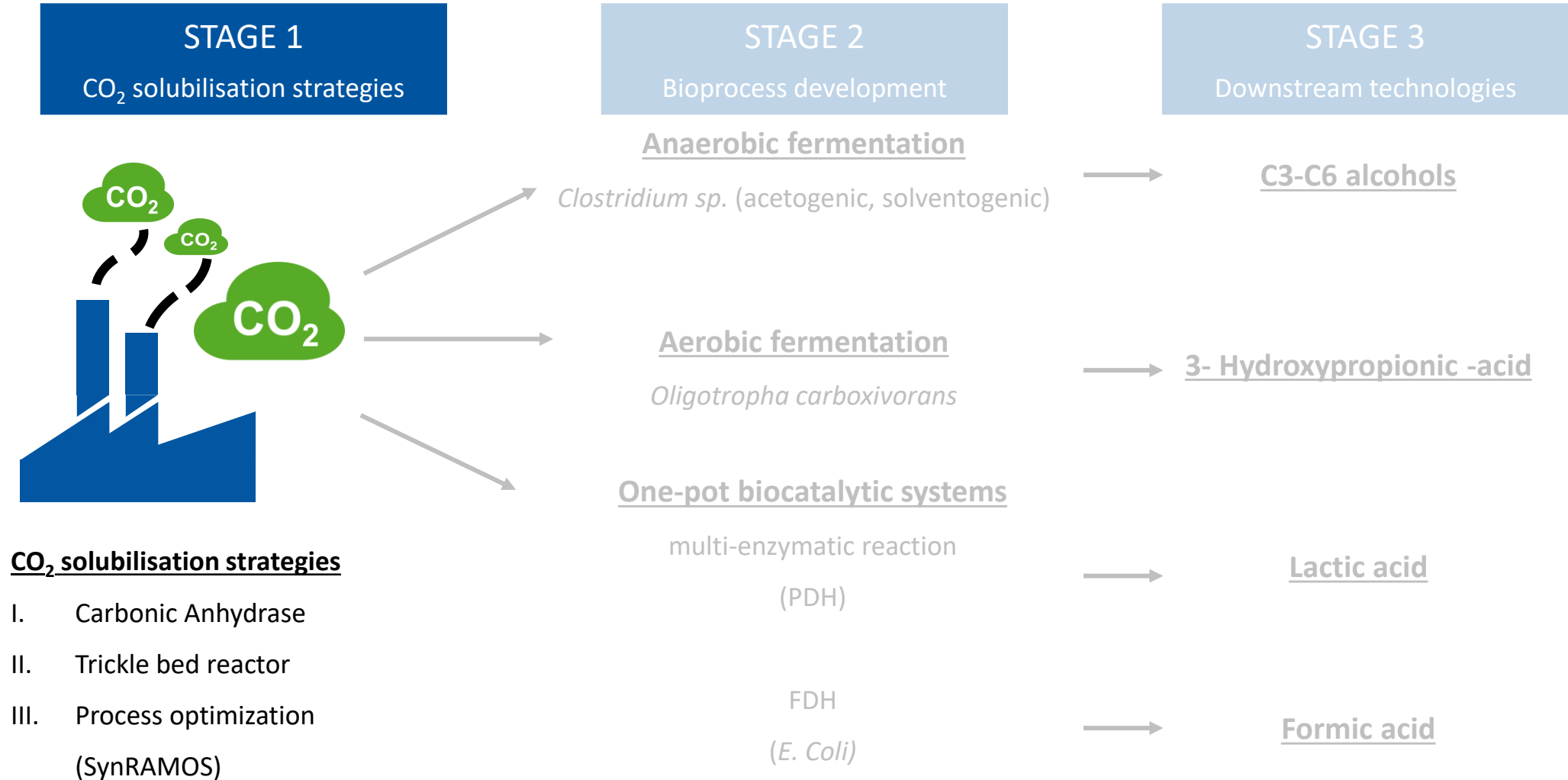
**Aline Hüser<sup>1</sup>, Marcel Mann<sup>1</sup>, Prof. Dr. Jochen Büchs<sup>1</sup>, Montse Bosch<sup>2</sup>, Aroa Rey Campa<sup>2</sup>, Rubén Rodríguez Alegre<sup>2</sup>, Mari Carmen Royo Reverter<sup>2</sup>**

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<sup>2</sup>LEITAT Technological Center, *Spain*

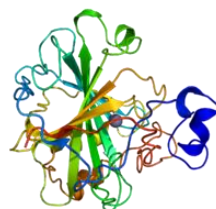


# Project Overview



# I. Carbonic Anhydrase: Benefits and Bottlenecks

## Human Carbonic Anhydrase II (hCAII)

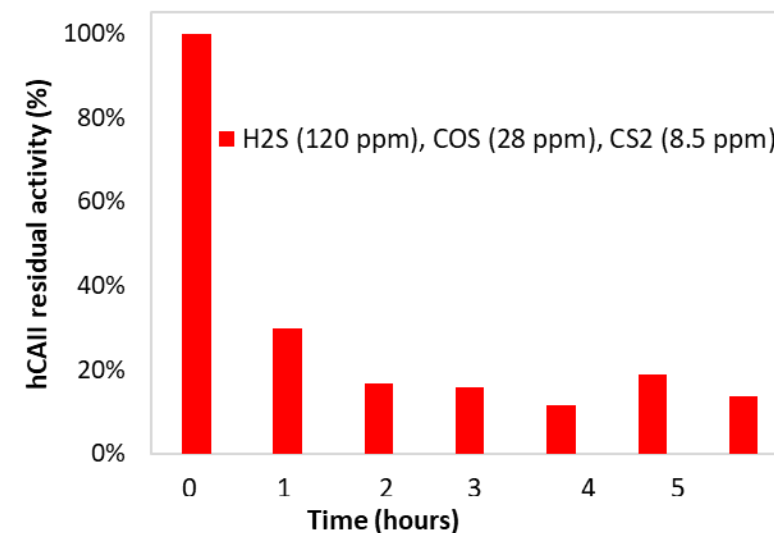
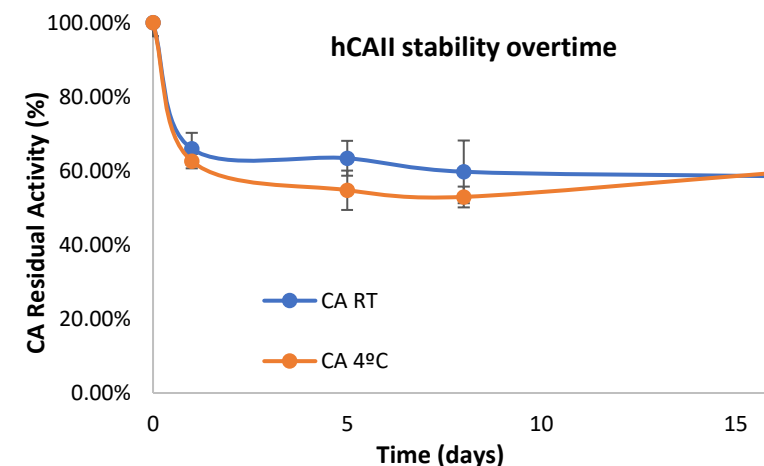


### Advantages

- ✓ High turnover rate ( $k_{\text{cat}} = 10^7 \text{ s}^{-1}$ )
- ✓ Mild operative conditions

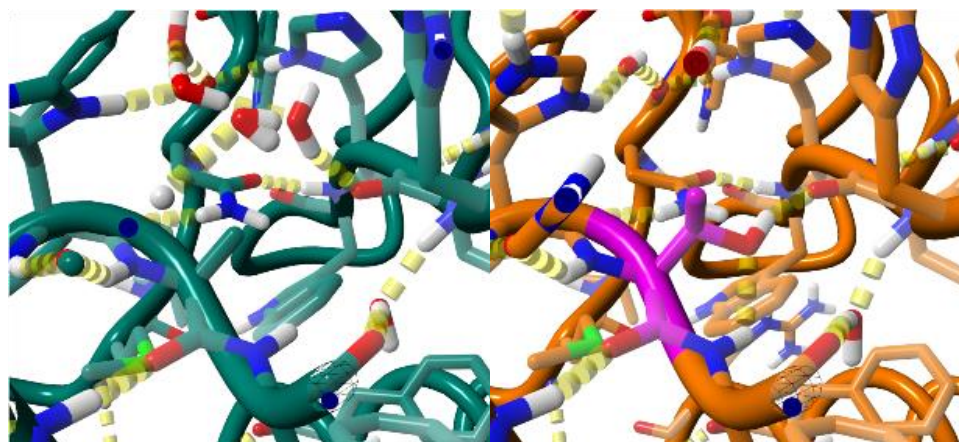
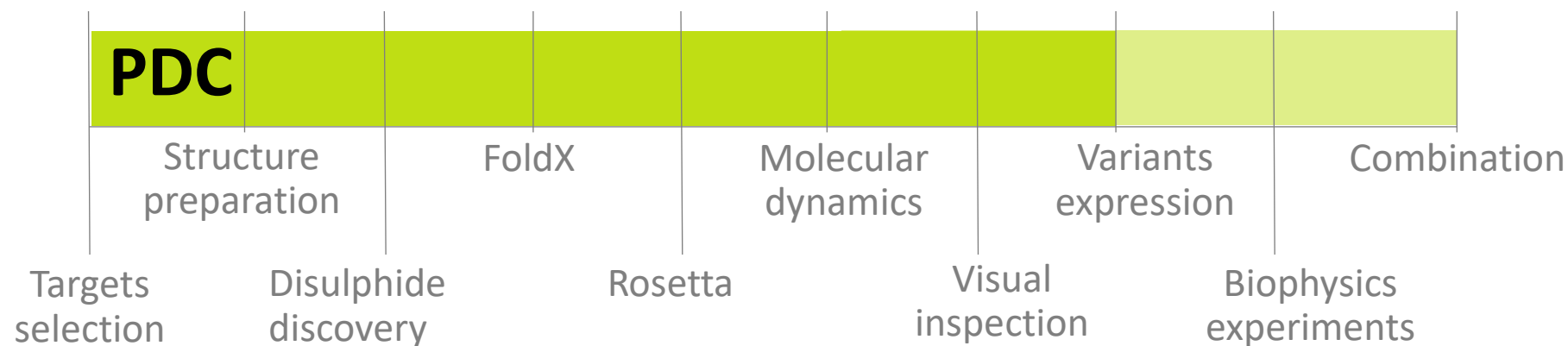
### Disadvantages

- ❑ Inhibition by sulphur-containing species
- ❑ Low Stability overtime



# Computational Strategy for Increased Stability

## Directed Evolution

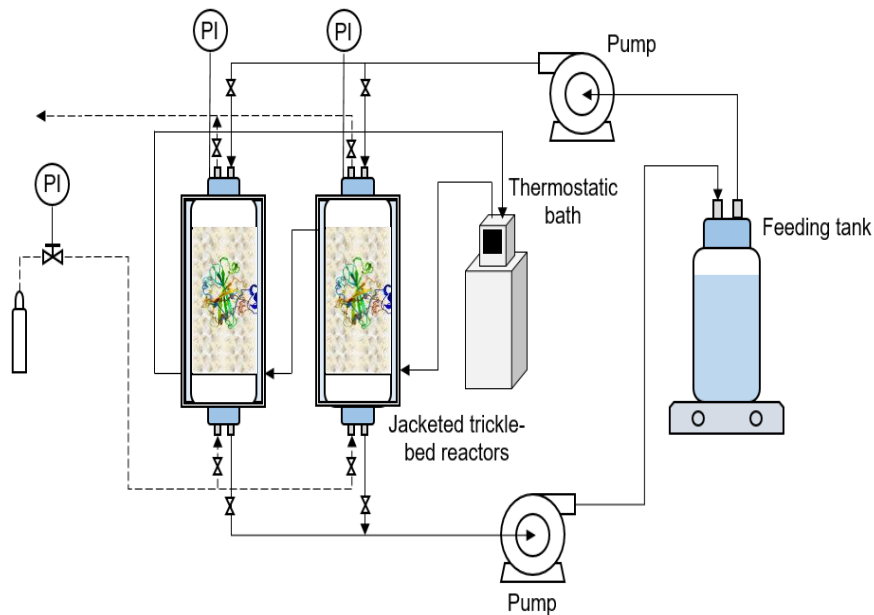


755 screen mutations

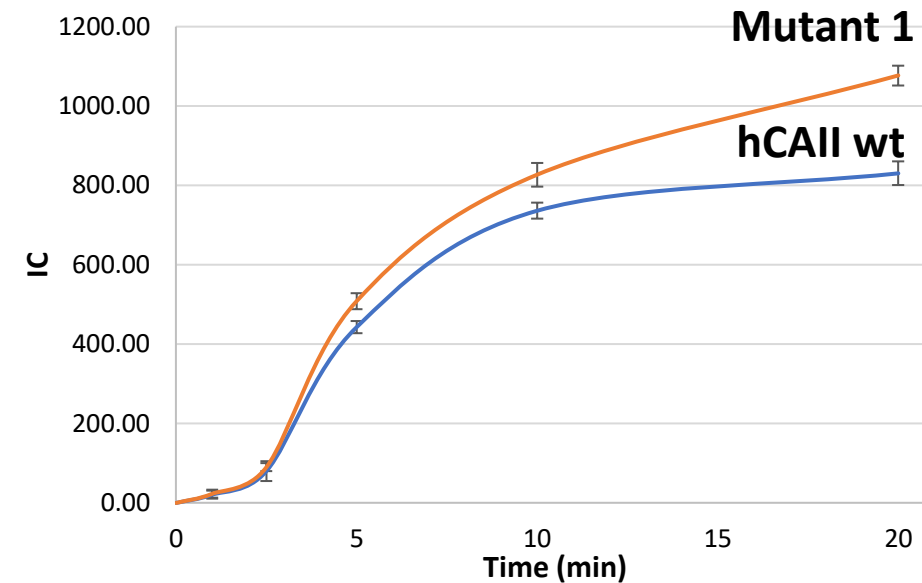
49 selected mutations

21 primer sets designed for Leitat

# CO<sub>2</sub> Solubilisation with Carbonic Anhydrases



Experimental set-up for CA solubilisation



- One of the mutants displays a higher activity (~1.5 fold) compared to hCAII wt but the same TM
- It shows **higher CO<sub>2</sub> solubilisation** with a maximum of 30% after 20 minutes

## II. Trickle-Bed Reactor

### Definition:

The Trickle-Bed Reactor (TBR) is a chemical reactor that uses the downward movement of a **liquid** and the downward (co-current) or upward (counter-current) movement of **gas** over a packed bed of (**catalyst**) particles.

### Critical parameters for the use in aerobic gas fermentation:

- Adhesion of bacteria *Cupriavidus necator* and biofilm formation
- Adaptability of the packing material inside the reactor
- Degradation suffered by the packing materials over time



# Advanced packing materials

The aim of this research is to study innovative, efficient, environmentally friendly and low-cost packing materials, by analyzing their characteristics, bio-adhesion properties and growth of bacteria

## POLYURETHANE FOAMS



*Hard PU foam*



*Soft PU foam*



*Polyisocyanurate (PIR) foam*

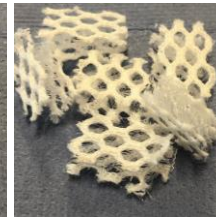
## FIBROUS MATERIALS (PES TEXTILES)



*Woven*



*Non-woven*



*3D*

## HARDWOOD CHIPS



*Beech wood*



*Eucalyptus wood*

## POLYPROPYLENE PELLETS



**REUSE AND WASTE  
MANAGEMENT:** Materials from  
industrial pre-consumer waste.



# TBR Conclusion

- BIOCON-CO<sub>2</sub> project has demonstrate the viability of TBR uses in CO<sub>2</sub> capture
- 10 different packing materials have been evaluated by means of surface characterization, behaviour inside the reactor and biofilm adhesion and growth
- It was determined the capability of packing materials to create biofilms by *C. necator*
- Plasma treatments on selected textiles:
  - Atmospheric plasma using gases to increase the roughness of the surface
  - PECVD plasma to functionalise the surface with permanent functional groups
  - Surface properties and biofilm formation can be tuned changing plasma conditions
- It is expected to have a publication soon with more results

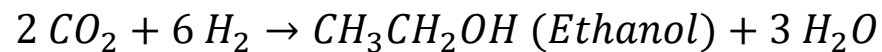
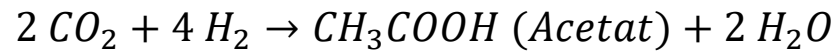




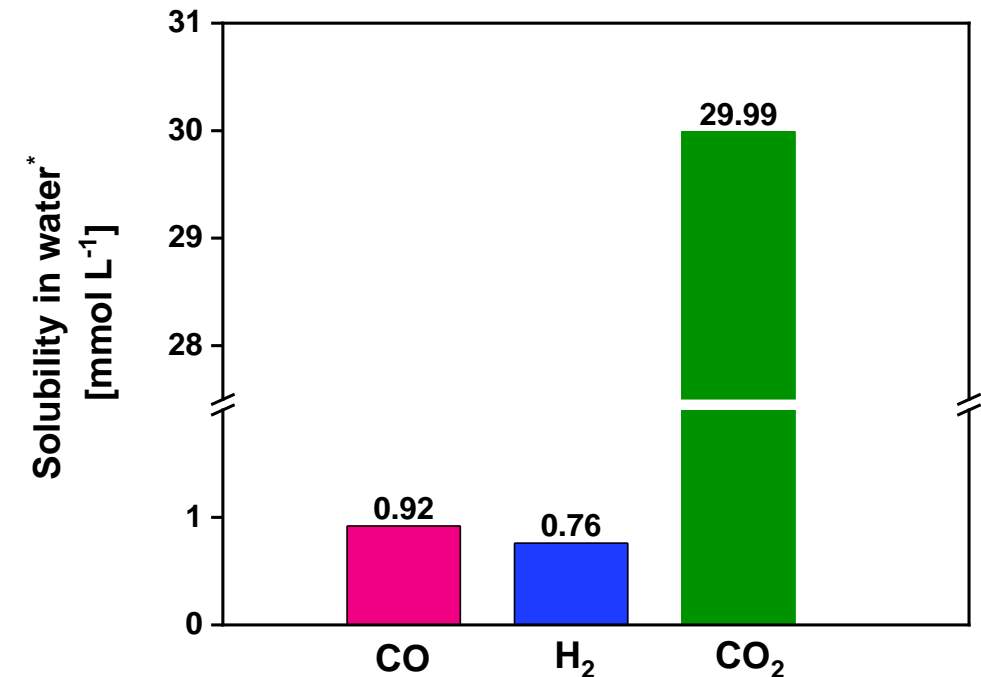
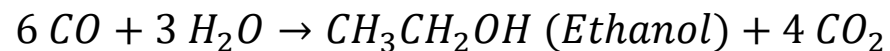
# III. Process optimization: Motivation

## Characteristics of syngas fermentation for CO<sub>2</sub> utilization

- Low substrate solubility for carbon monoxide and hydrogen
- CO<sub>2</sub> conversion depends on additional energy supply



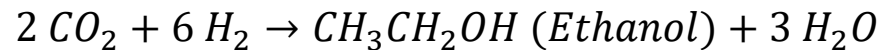
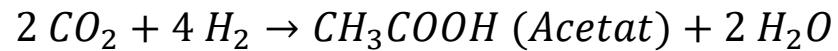
- CO conversion is energetically preferred and results in CO<sub>2</sub> production



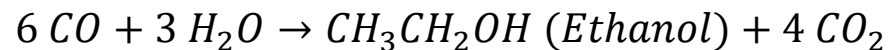
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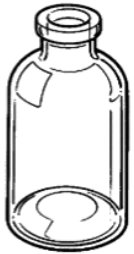
- CO conversion is energetically preferred and results in CO<sub>2</sub> production



- **Continuous gas supply** is necessary for sufficient growth
- **Monitoring CO<sub>2</sub> consumption/production** is highly important for process development

# State of the art process development for syngas fermentation

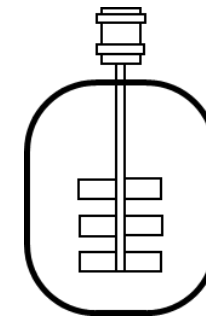
Serum bottle



Is there something to  
fill the gap?



Fermenter



No online monitoring

Limited gas supply

High throughput

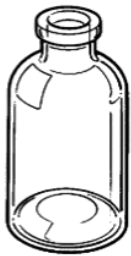
Online monitoring

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# State of the art process development for syngas fermentation

Serum bottle

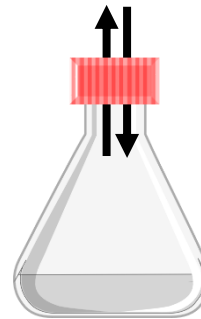


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Limited gas supply

High throughput

Gas shaker

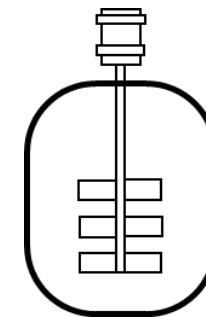


Online monitoring

Continuous gas supply

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Fermenter



Online monitoring

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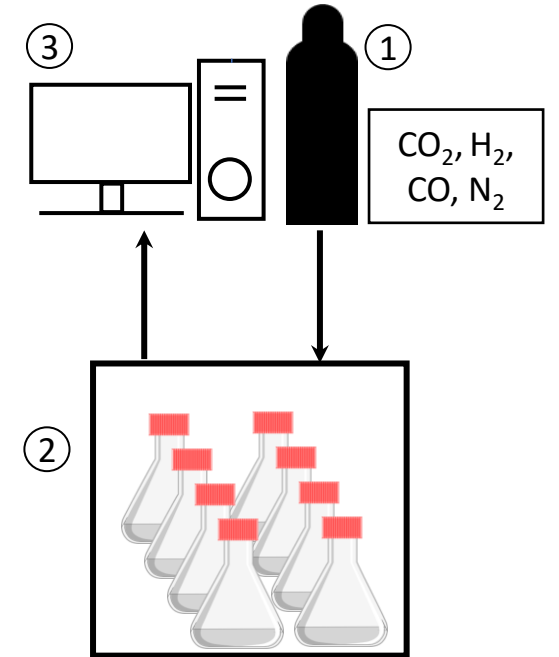
Limited throughput

# SynRAMOS - A device for gas fermentation in shake flasks

- Cultivation with gaseous carbon sources in up to 8 shake flasks
- Individually adjustable gas composition
- Safe to use with toxic and explosive gases (e.g. CO & H<sub>2</sub>)

## Measurement principle

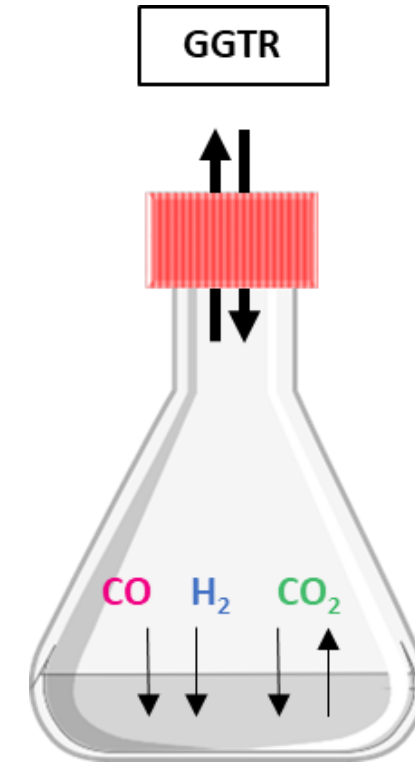
- Measurement of the headspace pressure in each flask
- Measurement of the CO<sub>2</sub> partial pressure via high range gas sensors



- (1) Gas supply
- (2) Shaking tablar with gasshaker setup
- (3) Computer

# Online Measurement of Gas Transfer Rates

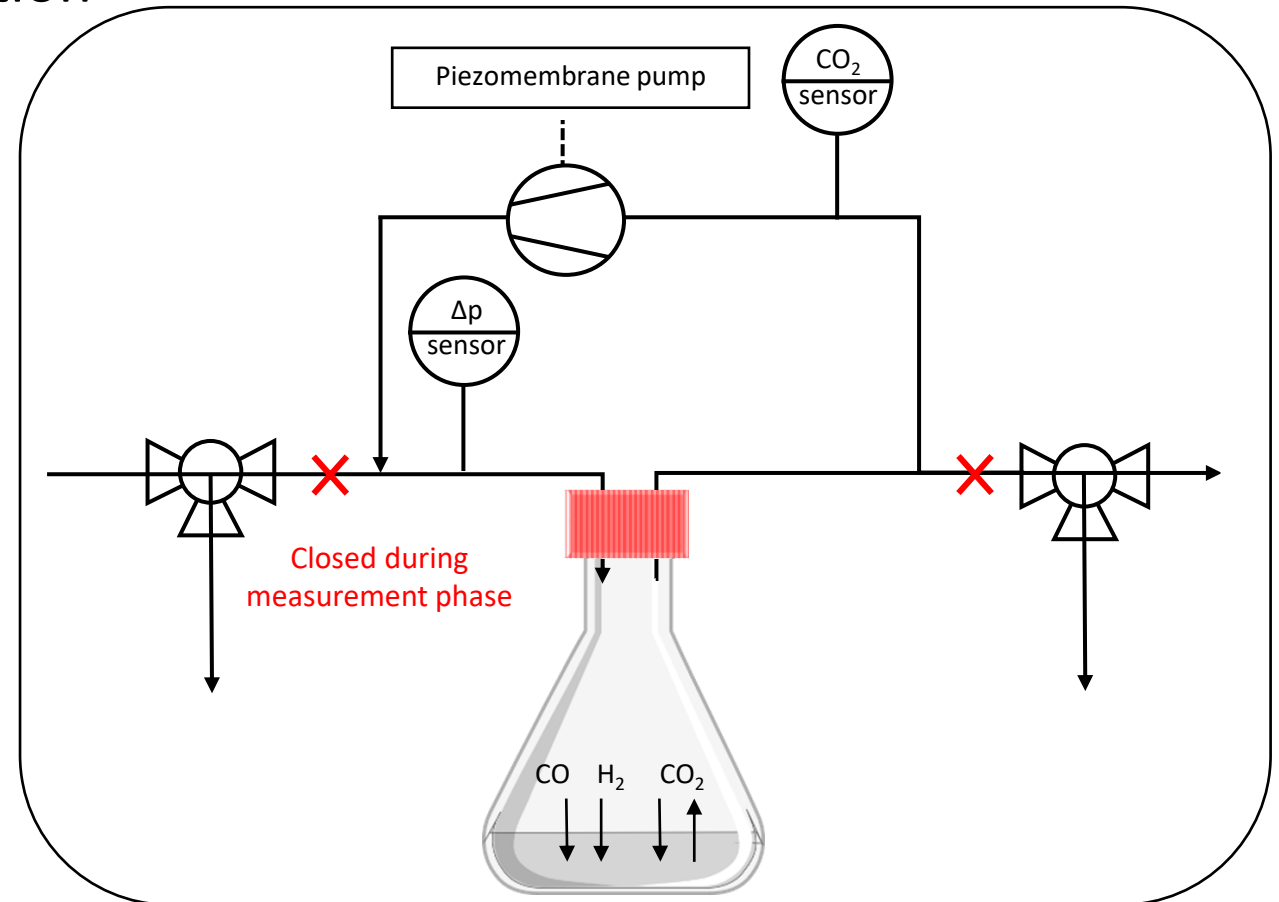
- Gross Gas Transfer Rate (**GGTR**) calculation
  - Determined via a pressure sensor
  - Represents the total gas transfer into and out of the liquid phase
- Carbon dioxide transfer rate (**CO<sub>2</sub>TR**) calculation
  - Measured via a CO<sub>2</sub>-Sensor
  - Represents the transfer of consumed or produced carbon dioxide



# Setup for the gas shaker

- Gross Gas Transfer Rate (GGTR) calculation

$$GGTR = \frac{dn_{total}}{dt} = \frac{\sum p_i \cdot V_{gas}}{V_{liquid} \cdot \Delta t \cdot R \cdot T}$$

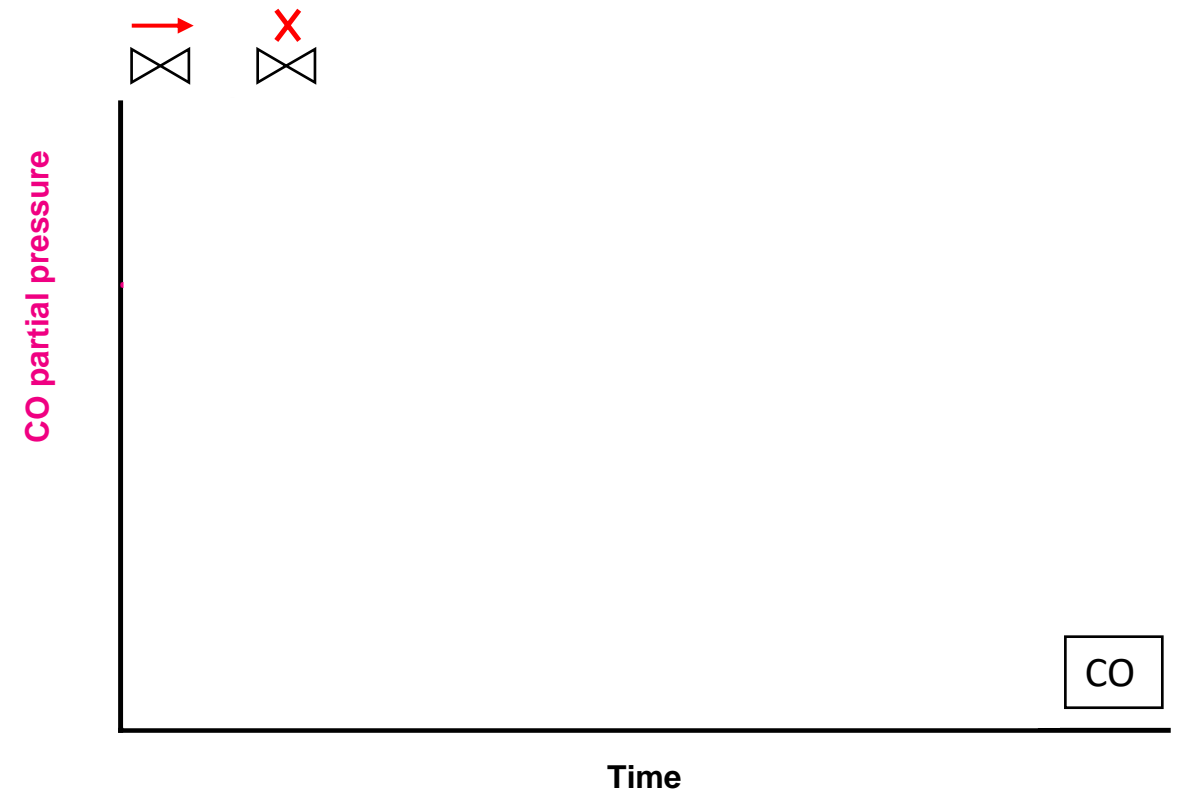
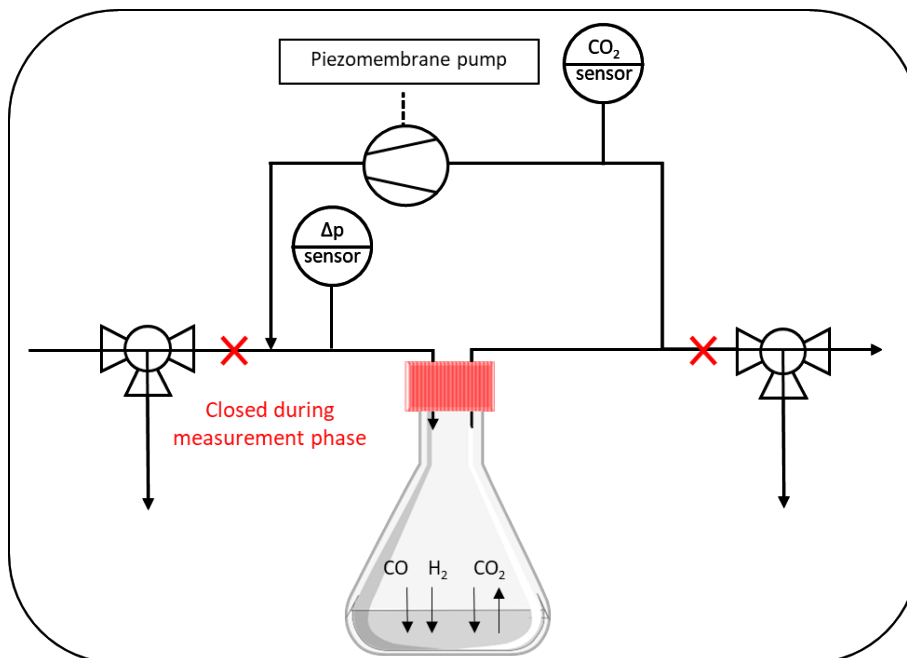




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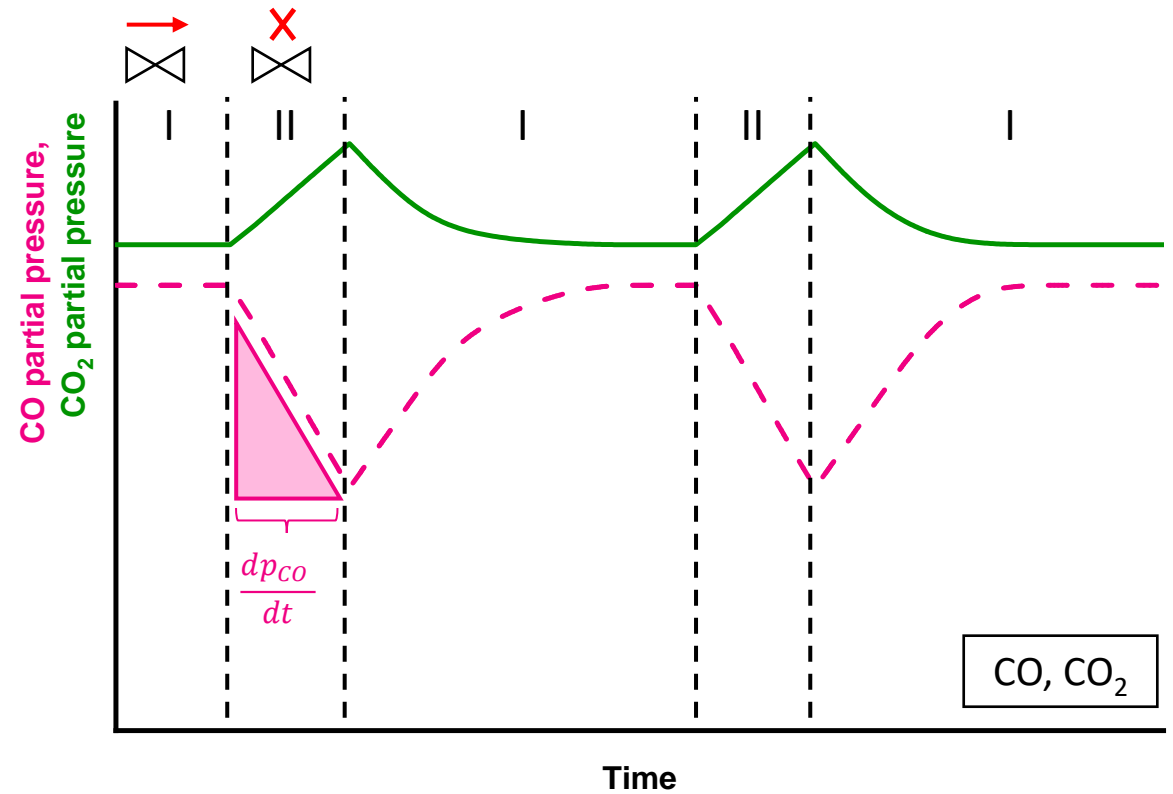
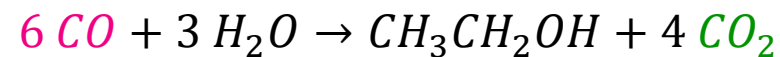
- I : Flush phase
- II: Measurement phase

# Measurement principle

- Gross Gas Transfer Rate (GGTR) calculation

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## CO as sole carbon source



I : Flush phase

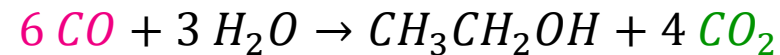
II: Measurement phase

# Measurement principle

- Gross Gas Transfer Rate (GGTR) calculation

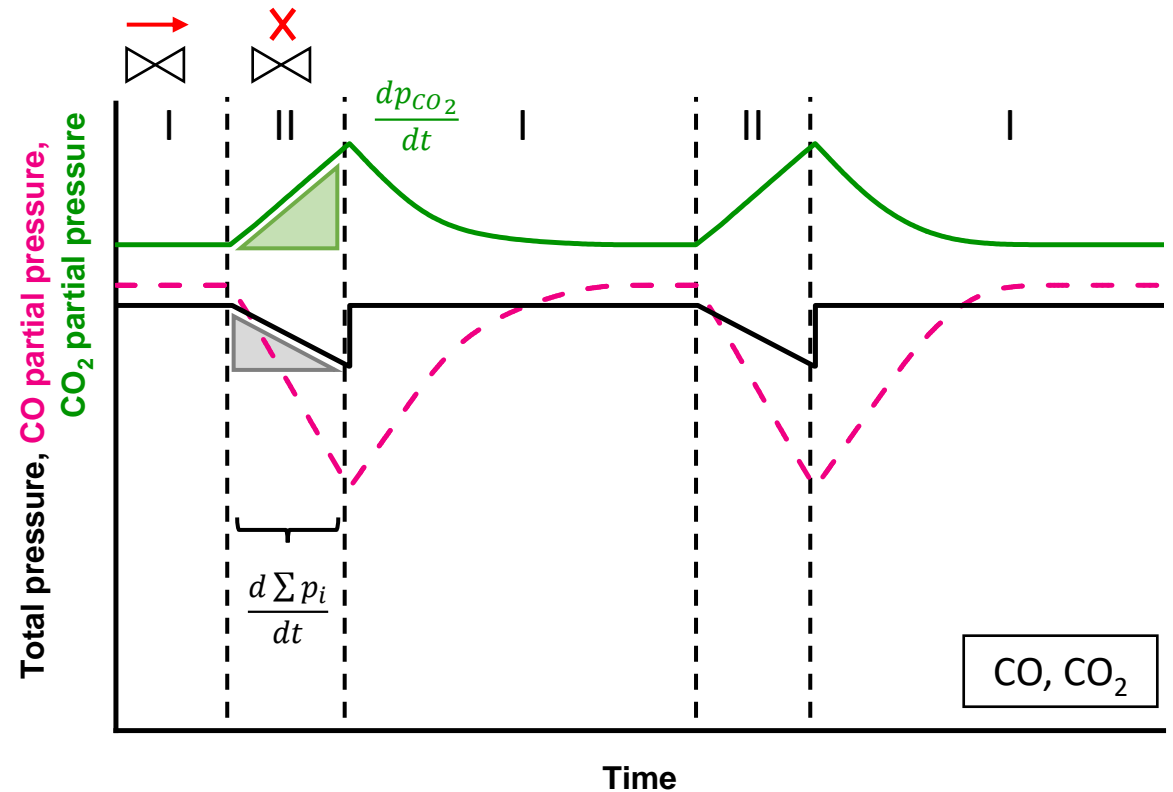
$$GGTR = \frac{dn_{total}}{dt} = \frac{\sum p_i \cdot V_{gas}}{V_{liquid} \cdot \Delta t \cdot R \cdot T}$$

## CO as sole carbon source



- Carbon Dioxide Transfer Rate (CO<sub>2</sub>TR) calculation

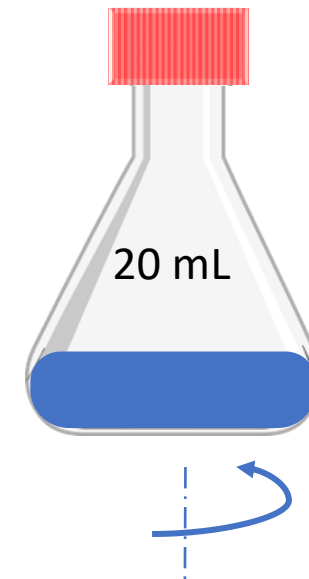
$$\text{CO}_2\text{TR}$$



I : Flush phase  
II: Measurement phase

# Case study - Gas fermentation using *C. ljungdahlii*

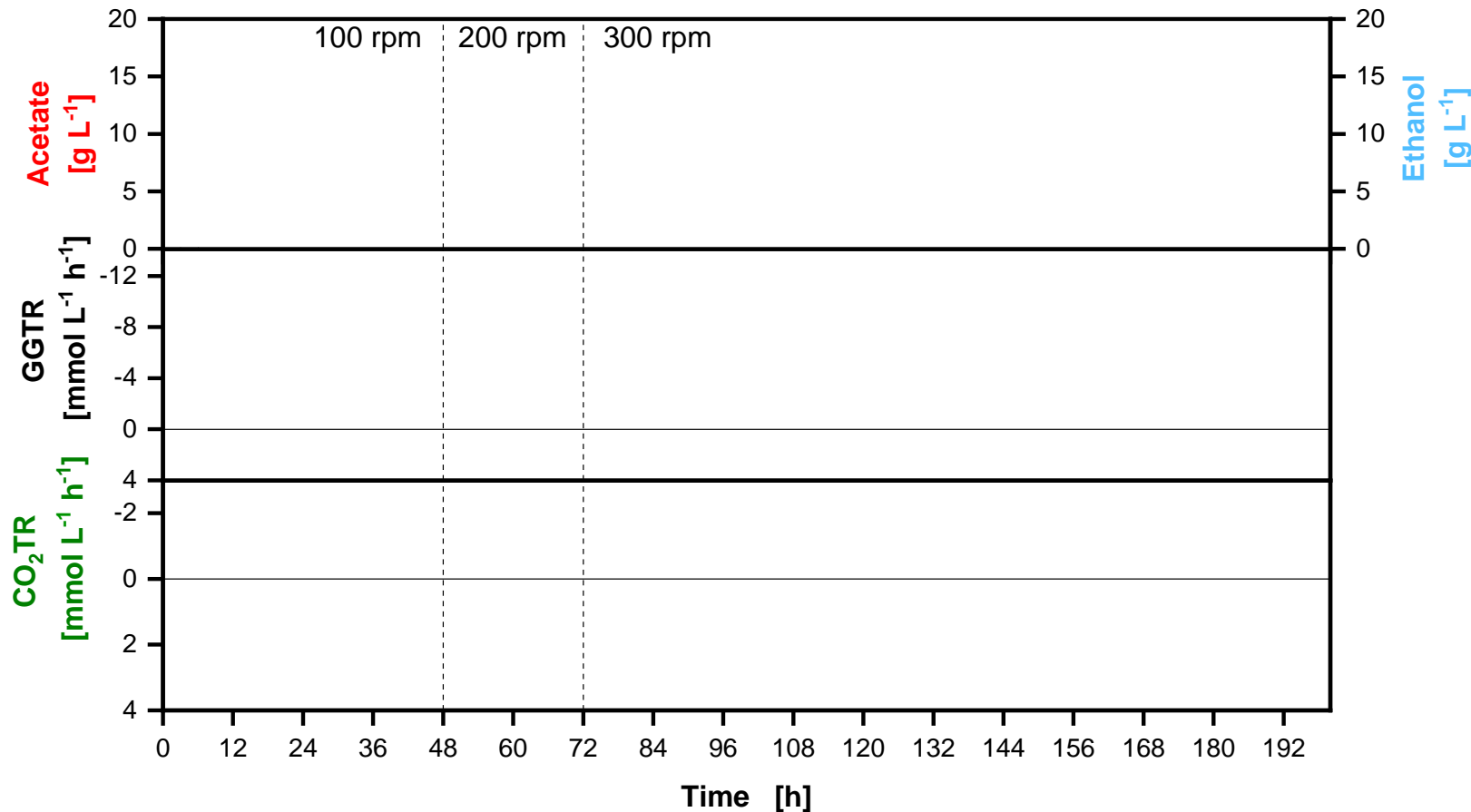
- Investigation of different gas transfer rates
    - Effect on gas consumption and **carbon dioxide conversion**
    - Enhanced product formation
  - Stepwise increase of shaking frequency
    - 100 rpm
    - 200 rpm
    - 300 rpm
- Higher shaking frequencies result in higher gas transfer rates



# Effect of increased gas transfer rates

*C. ljungdahlii* wildtype, ATCC media, T = 37°C, pH 7, 100 mmol BisTris, n = 100 - 300 rpm,

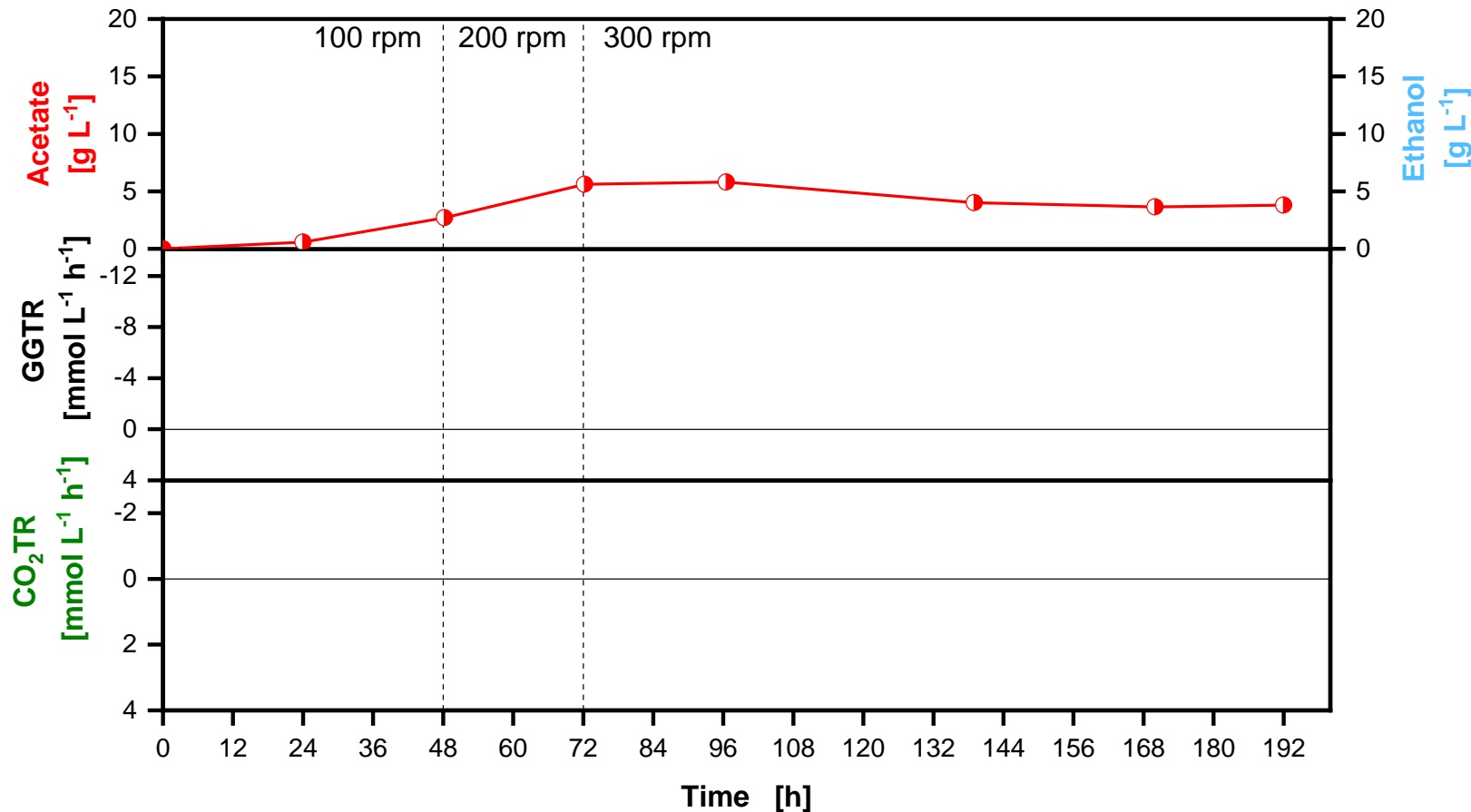
$d_0 = 50$  mm,  $\dot{V}_{GAS} = 5$  mL min<sup>-1</sup>, 10% CO / 20% CO<sub>2</sub> / 50% H<sub>2</sub> / 20% N<sub>2</sub>



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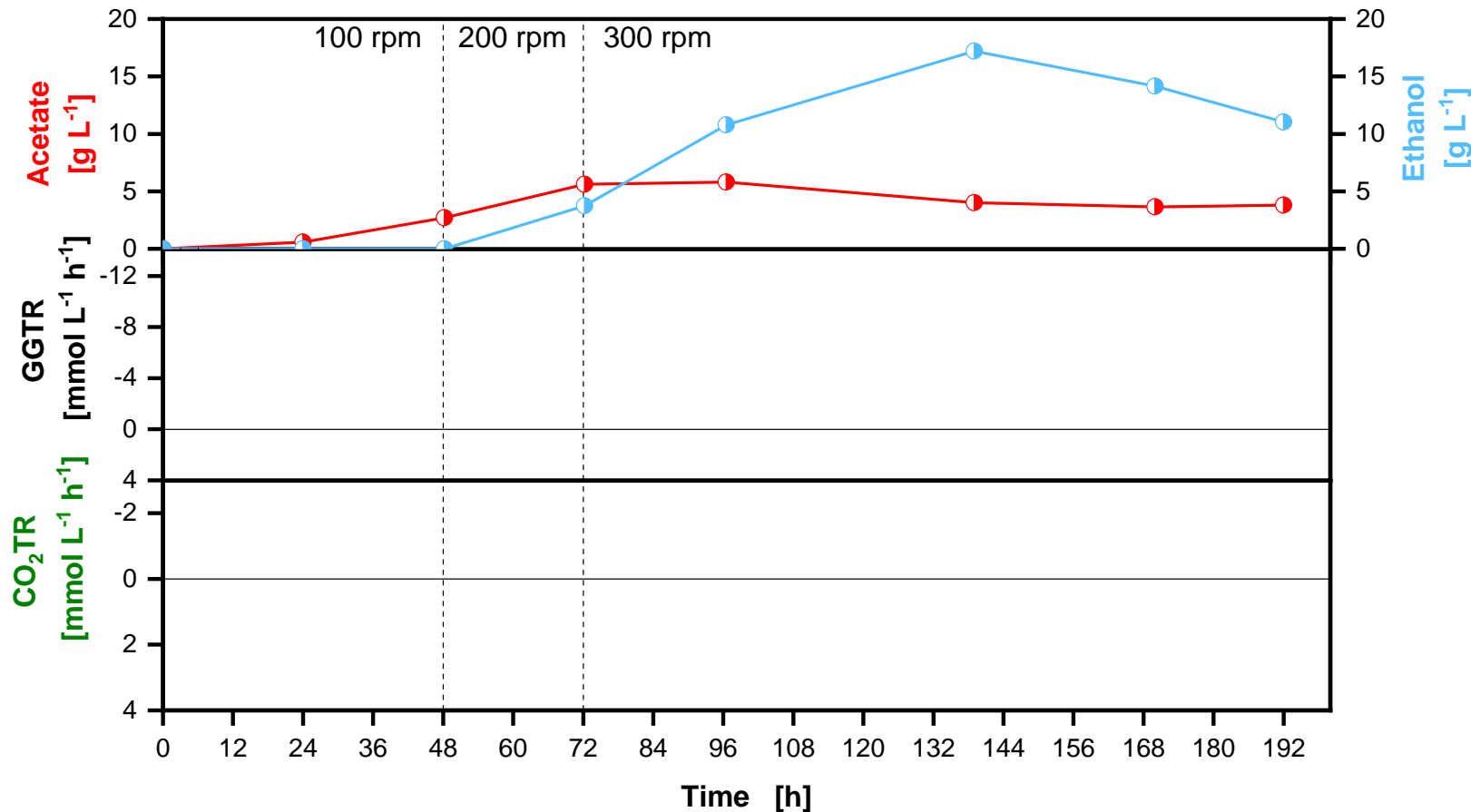


- Metabolic shift at 300 rpm

# Effect of increased gas transfer rates

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$d_0 = 50 \text{ mm}$ ,  $\dot{V}_{GAS} = 5 \text{ mL min}^{-1}$ , 10% CO / 20% CO<sub>2</sub> / 50% H<sub>2</sub> / 20% N<sub>2</sub>



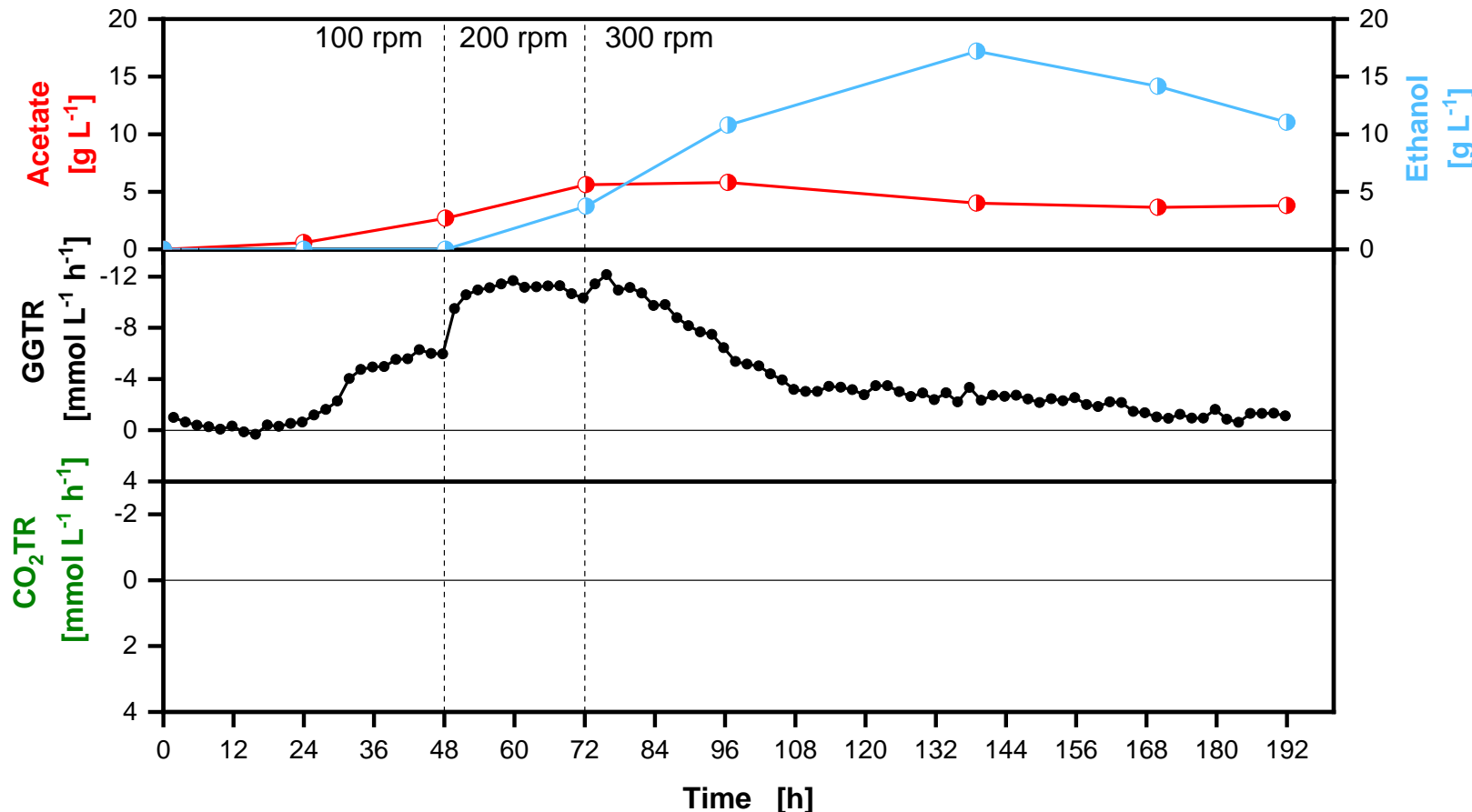
- Metabolic shift at 300 rpm
- Ethanol concentration of 15 g L<sup>-1</sup>



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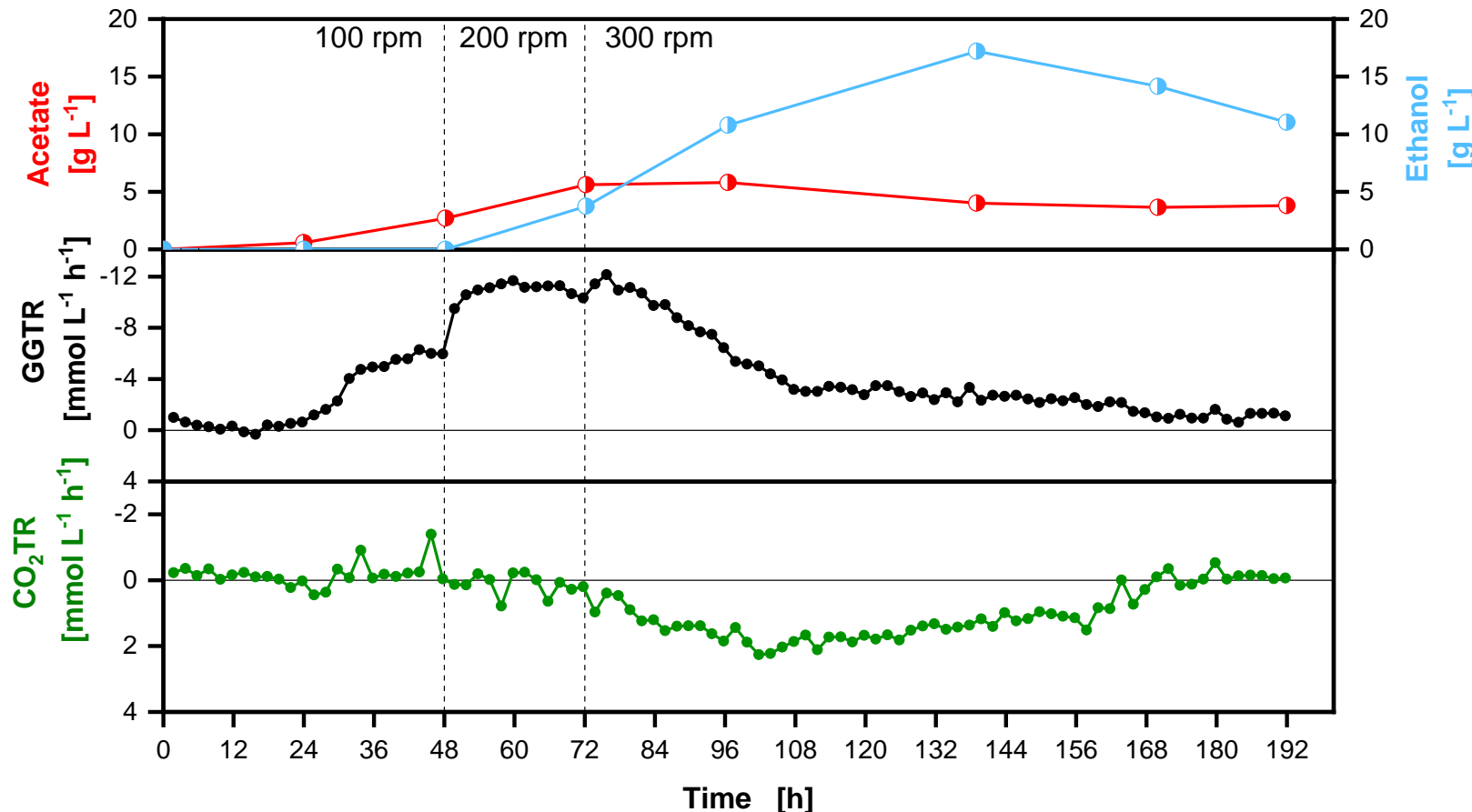


- Metabolic shift at 300 rpm
- Ethanol concentration of 15 g L<sup>-1</sup>
- Gas consumption increases with increasing shaking frequency
- GGTR drops after increase to 300 rpm

# Effect of increased gas transfer rates

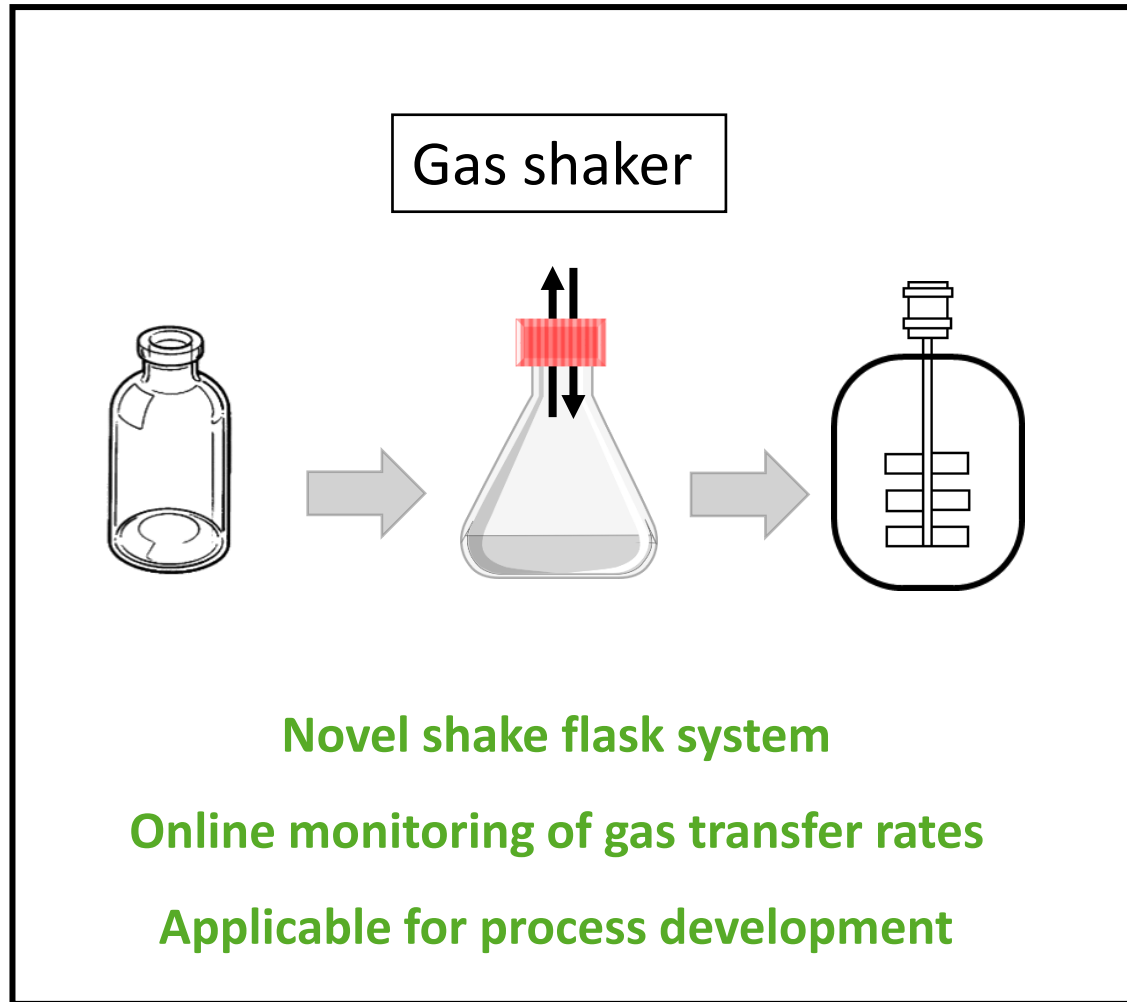
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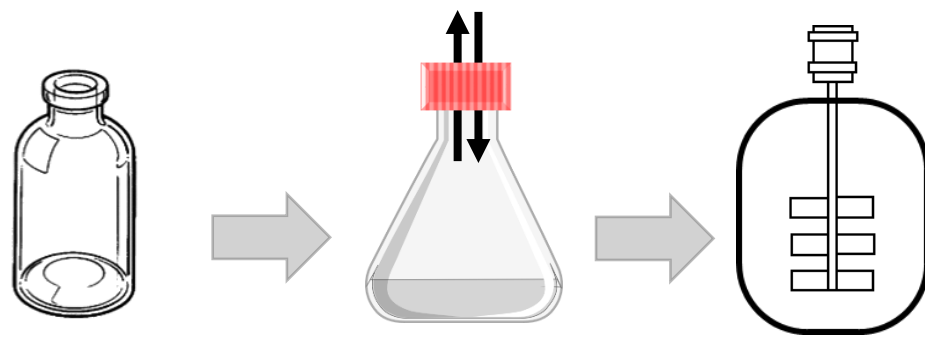


- Metabolic shift at 300 rpm
- Ethanol concentration of 15 g L<sup>-1</sup>
- Gas consumption increases with increasing shaking frequency
- GGTR drops after increase to 300 rpm
- CO<sub>2</sub>TR indicates excess CO<sub>2</sub> production after increase to 300 rpm
- ➔ Increasing gas transfer leads to hydrogenase inhibition

## III. Summary



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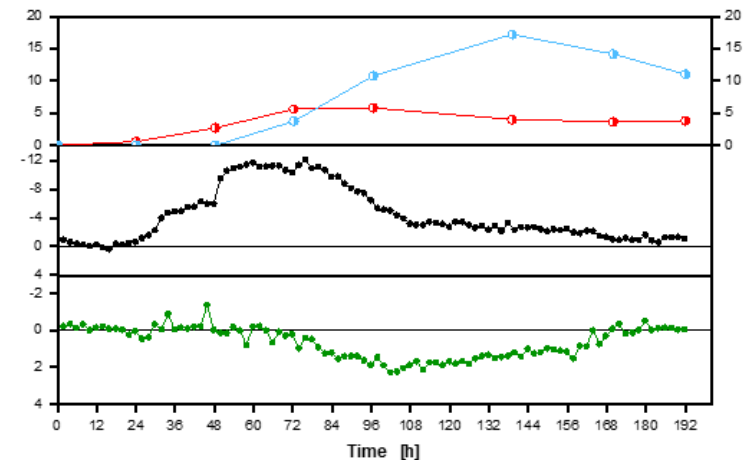
Gas shaker

Novel shake flask system

Online monitoring of gas transfer rates

Applicable for process development

Analyzing gas transfer rates



Online measurement of GGTR and CO<sub>2</sub>TR  
enables deeper insights into small scale  
gas fermentation processes.



# Thank you

**Aline Hüser, Marcel Mann, Prof. Dr. Jochen Büchs,  
Montse Bosch, Aroa Rey Campa, Rubén Rodríguez Alegre,  
Mari Carmen Royo Reverter**



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